

Name _____

Balancing Equation Practice

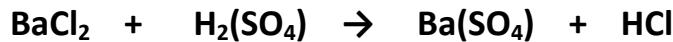
Part A: Identify the following parts of each chemical formula by circling the subscripts and drawing a square around the coefficients.

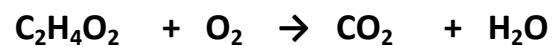


Part B: List the symbols for the atoms in each formula and give the number of each.



Part C: Balance each of the following equations following the procedure described in class. Be sure to show your work.





Write out the equation and then balance the equation.

Carbon + Hydrogen \rightarrow Carbon Tetrahydride

Sodium + Chloride \rightarrow Sodium Chloride